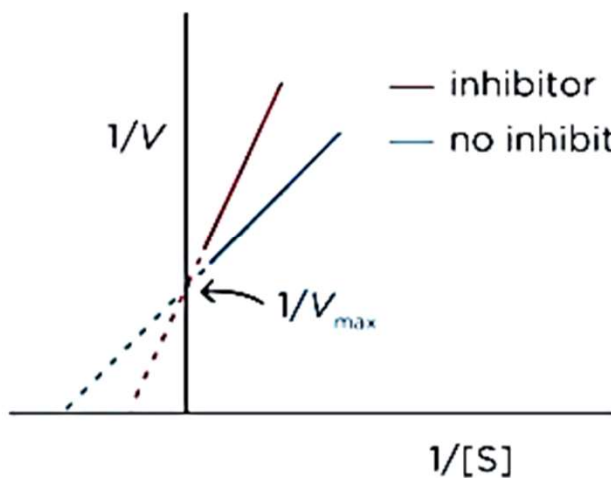


# MATHEMATICS

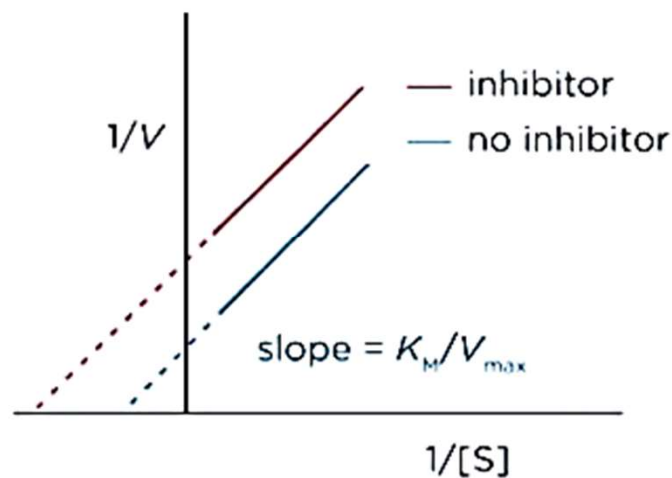
 @DiligentEdu

# ENZYME INHIBITION



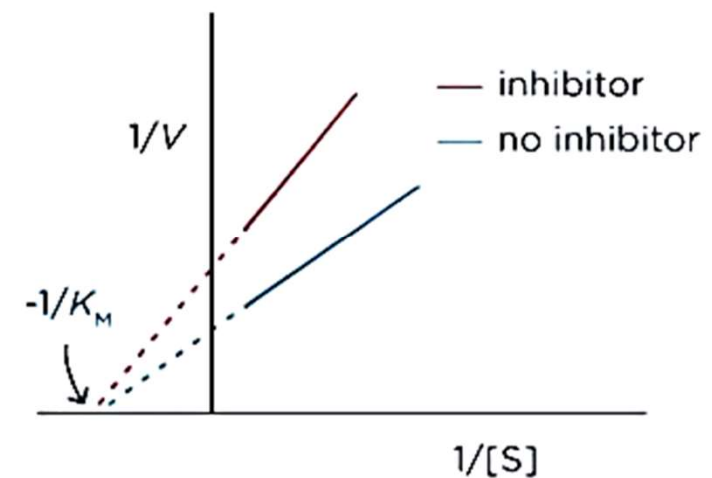
**Competitive inhibition**

$K_M$  increased  
 $V_{\max}$  unaffected



**Uncompetitive inhibition**

$K_M$  reduced  
 $V_{\max}$  reduced



**Noncompetitive inhibition**

$K_M$  unaffected  
 $V_{\max}$  reduced

source: Biochemistry, epathshala, A Gateway to All Post Graduate Courses, MHRD

# ENZYME INHIBITION

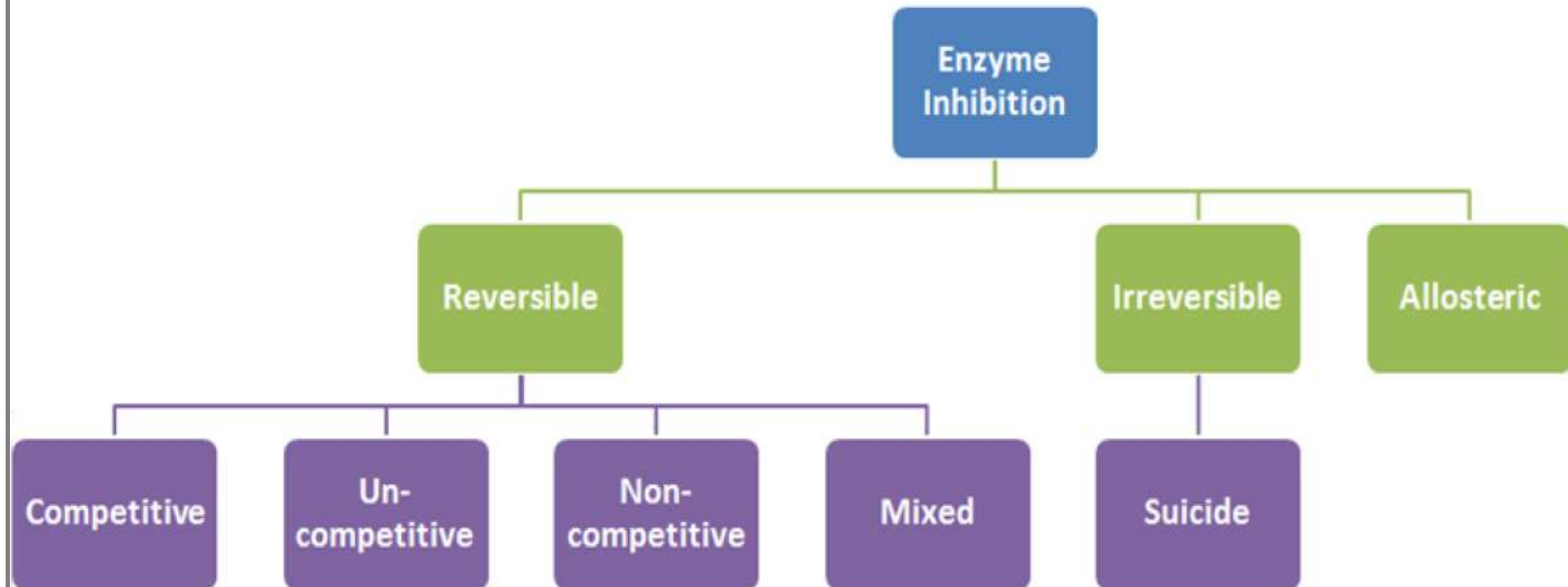
## ■ Introduction – Enzyme Inhibition

An enzyme catalyzed reaction can be hindered or reduced by a number of substances. Some others like **urea** are known as *denaturants*, being *non-specific* in their mode of action.

*But if* any compound act in a fairly *specific way* in inhibiting the catalysis of a particular enzyme they are called **inhibitors**.

# ENZYME INHIBITION

## ■ Enzyme Inhibition - *Types*



# ENZYME INHIBITION

## ■ Enzyme Inhibition - *Types*

The loss in activity can either be of two types -  
(1) **Reversible** - where the activity *can be restored* by the *removal* of the inhibiting compound. *It's temporary.*

(2) **Irreversible** - where the loss of activity *cannot be recovered within the stipulated time* of interest. *It is permanent. ...*

# ENZYME INHIBITION

## ■ Enzyme Inhibition - *Types*

... Irreversible inhibition behave as *time dependent loss of enzyme concentration* with lowered  $V_{\max}$  or *incomplete in activation* with *time dependent change* in both  $K_m$  and  $V_{\max}$ .

...

# ENZYME INHIBITION

## ■ Enzyme Inhibition - *Types*

... Heavy metal ions like mercury, lead etc. *cause irreversible inhibition*, which bind strongly to the *amino acid backbone* termed as “**suicide inhibition**”.

# ENZYME INHIBITION

## ▪ Studying Enzyme Inhibition Mechanisms

There are several reasons behind the need for *studying enzyme inhibition mechanisms*.

They are –

- Exploring potential mechanisms in multi-substrate reactions.
- Studying the relative binding affinity of competitive inhibitors to the enzyme active ...

# ENZYME INHIBITION

## ■ Studying Enzyme Inhibition Mechanisms

There are several reasons behind the need for *studying enzyme inhibition mechanisms*.

They are –

... site, in the absence of 3-D structure information.

- For understanding various control mechanisms - how the balance of protease ...



# ENZYME INHIBITION

## ■ Studying Enzyme Inhibition Mechanisms

There are several reasons behind the need for *studying enzyme inhibition mechanisms*.

They are –

... enzymes and their inhibitors in tissues achieve **homeostasis**.

- For various **commercial applications** like *pesticide, insecticide, weed-killers, pharmaceutical compounds* like *drugs* etc.

# ENZYME INHIBITION

## ■ Reversible Inhibition

In this type of inhibition, the *hindrance is temporary* and thus **noncovalent interactions** like **hydrogen bonds**, **ionic bonds** or **hydrophobic bonds** form between inhibitors and the enzyme. Even though these are *weak bonds*, multiple such bonds cause strong and specific binding. ...

# ENZYME INHIBITION

## ■ Reversible Inhibition

... Despite absence of any chemical reactions, the inhibitors can easily remove or exchange by dilution or dialysis. After removing the inhibitor, **enzyme** can be fully **restored** in reversible inhibition. *Equilibrium* is established between *free inhibitor* and *enzyme-inhibitor [EI] complex* (Figure 2).

# ENZYME INHIBITION

## ■ Reversible Inhibition

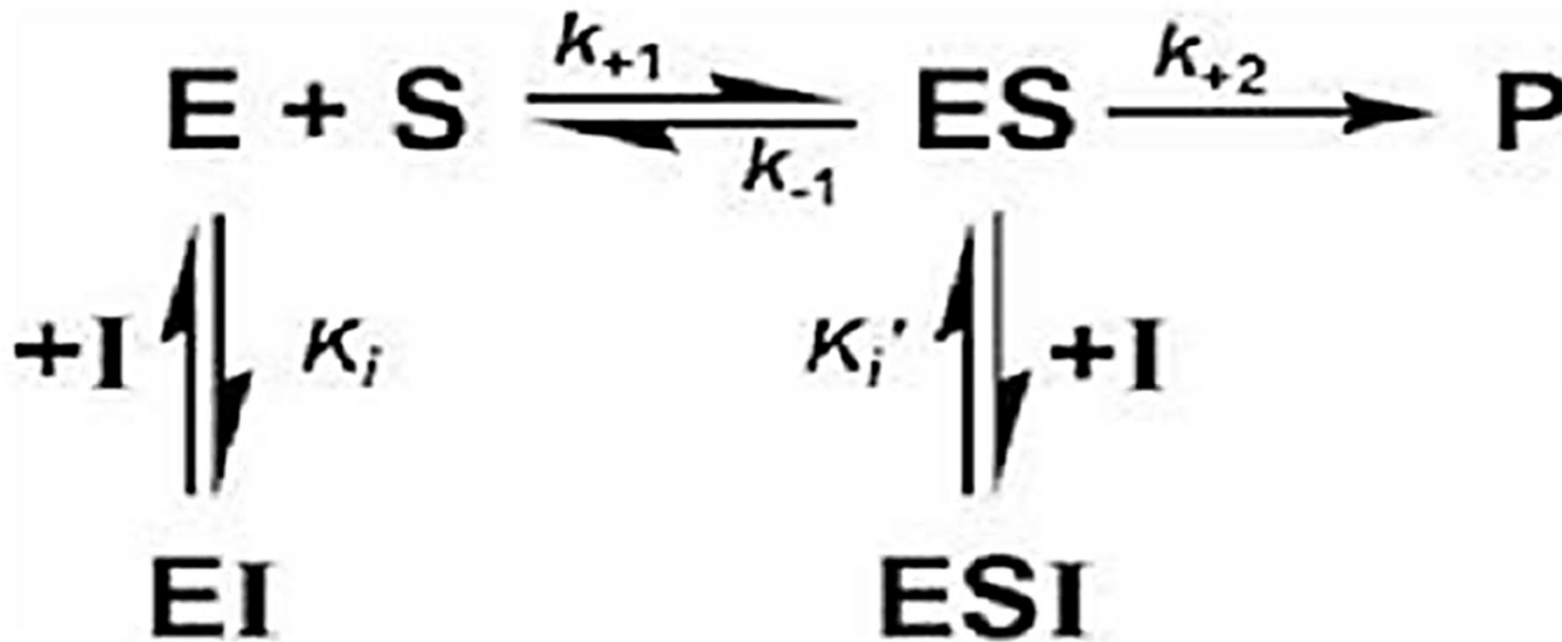
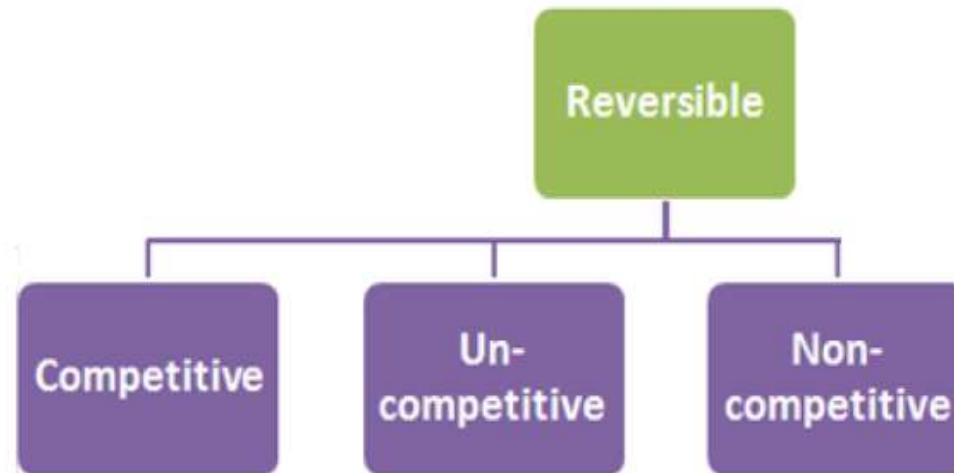


Figure 2. Mechanism of reversible inhibition

# ENZYME INHIBITION

## ■ Reversible Inhibition - Types

Reversible inhibitions are of different types. The classification is **based** according to the effect of **varying the concentration of the enzyme's substrate on the inhibitor.**



# ENZYME INHIBITION

## ■ Competitive Inhibition

In this type of reversible inhibition, both the substrate and its inhibitor *cannot bind* to the enzyme *at the same time* to the allosteric / active site. This normally occurs *due to* the structural similarity of substrate and the inhibitor, which results with *affinity* for the active site. ...

# ENZYME INHIBITION

## ■ Competitive Inhibition

... The inhibition *can be recovered by* the presence of **high concentration of substrate**, outcoming the competing inhibitor.  $V_{\max}$  of the reaction is **unchanged**, while  $k_d$ , the **dissociation constant** is apparently **increased**. Competitive inhibitors can also be used to find the enzyme active site.

# ENZYME INHIBITION

## ▪ Competitive Inhibition

Example-1: N-(Phosphonoacetyl)-L-asparate also known as **PALA** is a **competitive inhibitor** for *Aspartate Transcarbamoylase*.

Example-2: **Malonate** is a **competitive inhibitor** of enzyme *succinate dehydrogenase*, and *competes with succinate*.



# ENZYME INHIBITION

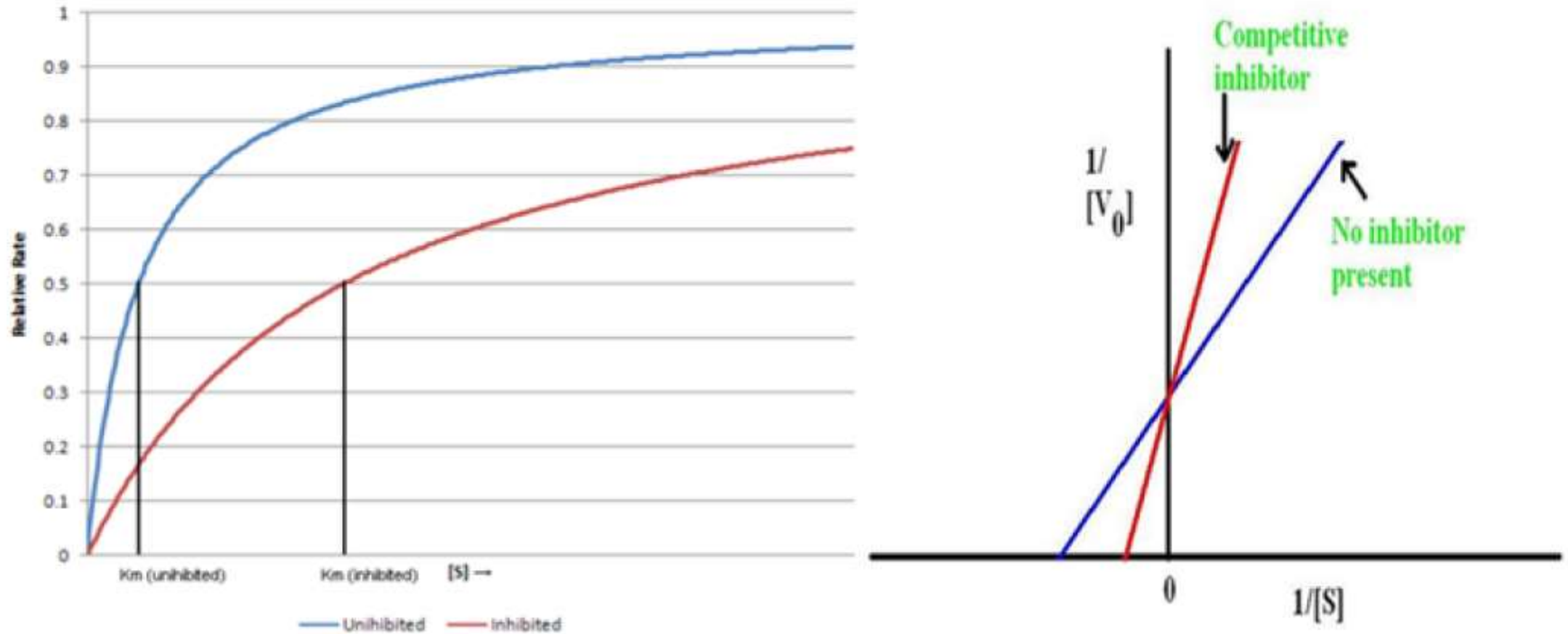


Figure 3. Kinetics of competitive inhibition [DiligentEdu](#)®

# ENZYME INHIBITION

Table 1. Clinical use of competitive inhibition

Drugs	Target Enzyme	Therapeutic use
<b>STATINS</b> , Atorvastatin Simvastatin	HMG CoA reductase	Involved in the reduction of plasma cholesterol level- Anti-hyperlipidemic agents
<b>Allopurinol</b>	Xanthine oxidase	Used in case of prevention of gout attacks
<b>Methotrexate</b>	Dihydrofolate reductase	Used as cancer treatment drug
<b>Captopril, Enalapril</b>	Angiotensin converting enzyme	Treatment of high blood pressure
<b>Discoumarol</b>	Vit K-epoxide reductase	Used as an anti-coagulant

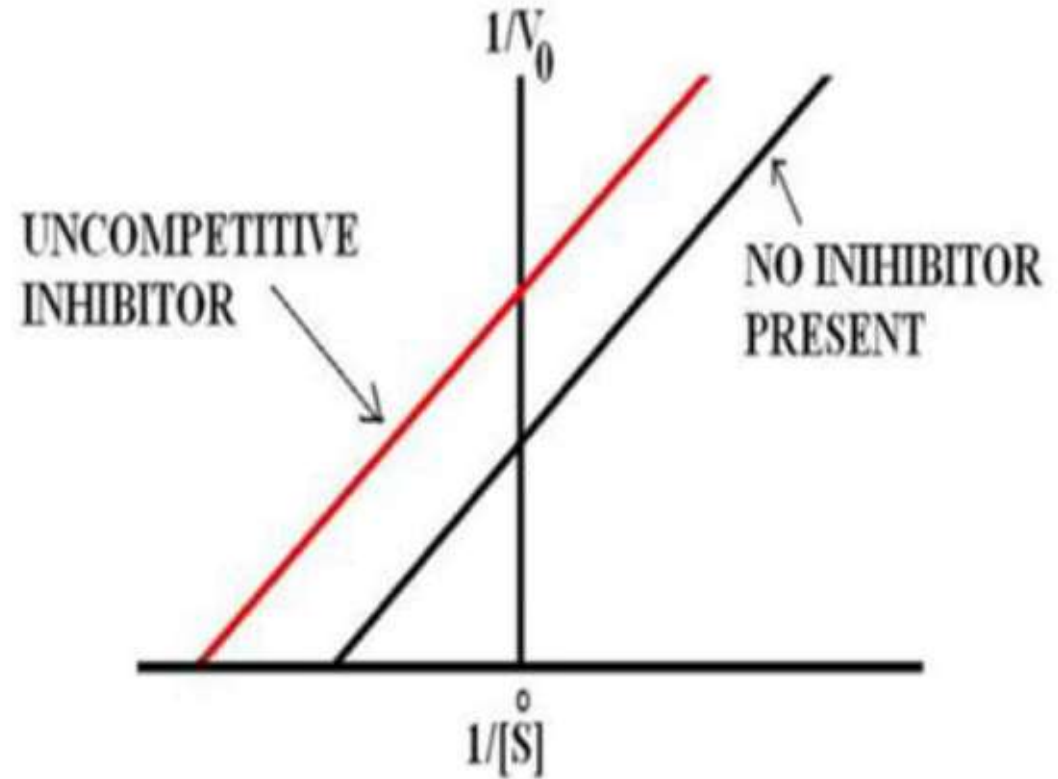
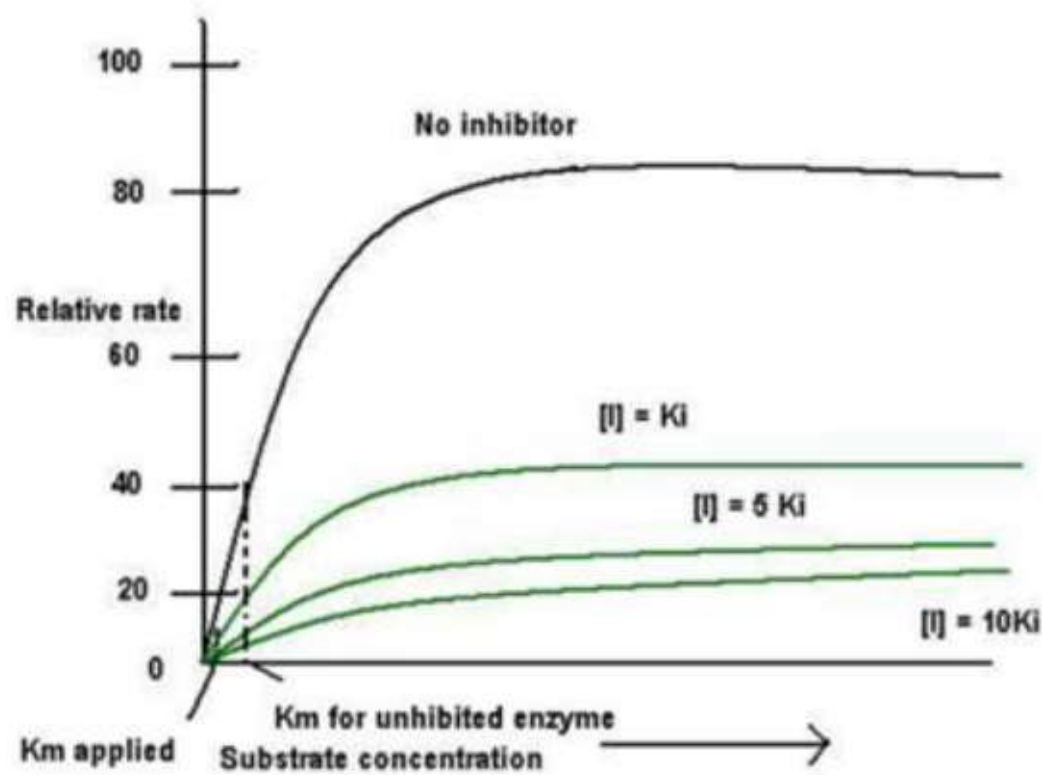
# ENZYME INHIBITION

## ■ Uncompetitive Inhibition

It's an **anti-competitive inhibition**; where the inhibitor binds **only** to the substrate-enzyme complex. According to its kinetics,  $V_{\max}$  and  $K_m$  decrease. This type of inhibition works best in case of *high concentration* of the *substrate*. The substrate and the uncompetitive inhibitor **does not resemble** each other.

**Example** - *Lithium and phosphoinositide cycle.* 

# ENZYME INHIBITION



**Figure 4.** Kinetics of un-competitive inhibition

# ENZYME INHIBITION

## ■ Non-Competitive Inhibition

A non-competitive inhibitor is one which reacts with enzyme-substrate or [ES] complex. It does not affect the binding of the substrate, but slows down the reaction rate for formation of the enzyme-product [EP] complex. ...

# ENZYME INHIBITION

## ■ Non-Competitive Inhibition

... The only factor on which the extent of hindrance or inhibition **depends** is the **inhibitor concentration**. There will be a *decrease* in  $V_{\max}$  but  $K_m$  will remain the *same*.

**Example:** Alanine non competitively inhibits the enzyme **pyruvate kinase**.

# ENZYME INHIBITION

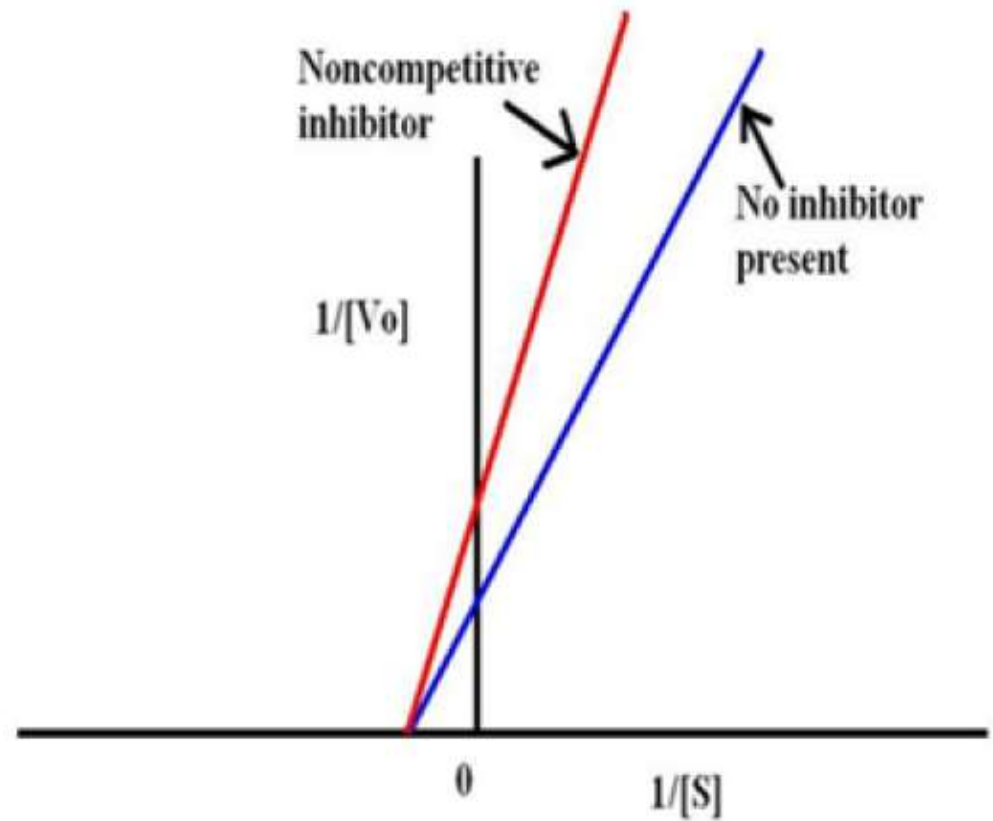
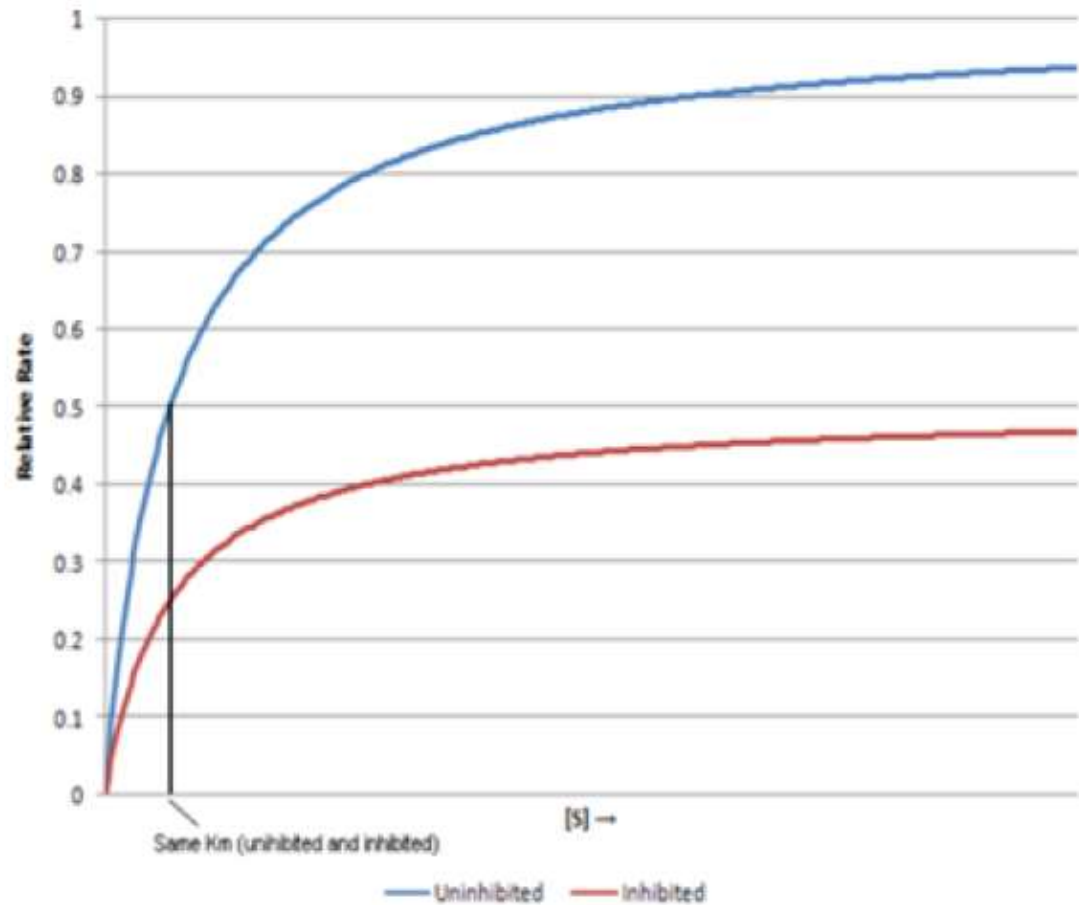


Figure 5. Kinetics of non-competitive inhibition

# ENZYME INHIBITION

**Table 2. Examples of common non-competitive inhibitors**

Inhibitor	Enzyme inhibited
Heavy metals – $\text{Ag}^{2+}$ , $\text{Hg}^{2+}$ , $\text{Pb}^{2+}$	Heavy metals bind with cysteinyl SH group of enzyme
Pepstatin	Pepsin
Soybean trypsin inhibitor	Trypsin
Ethanol/narcotic drugs	Acid phosphatase



# ENZYME INHIBITION

## ■ Mixed Inhibition

In this type of inhibitor, **inhibitor** is capable of **binding to both *free enzyme* as well as *enzyme-substrate complex***. In this case,  $V_{\max}$  and  $K_{\max}$  varies. **Mixed inhibitor binds to the *allosteric site***. This type of inhibition **cannot overcome by increasing substrate concentration S**, but can be reduced. ...

# ENZYME INHIBITION

## ■ Mixed Inhibition

... The inhibitor binding to the *allosteric site* *changes the* structural confirmation to reduce the *affinity* of the substrate.

**Example:** Mixed inhibition is observed on case of *oxidoreductase activity* of *Xanthine oxidase* by  $\text{Pd}^{2+}$  ion.

# ENZYME INHIBITION

$$\frac{1}{v} = \frac{k_s}{v_{\max}} \left(1 + \frac{[I]}{k_i}\right) \frac{1}{[S]} + \frac{1}{v_{\max}} \left(1 + \frac{[I]}{\alpha k_i}\right)$$

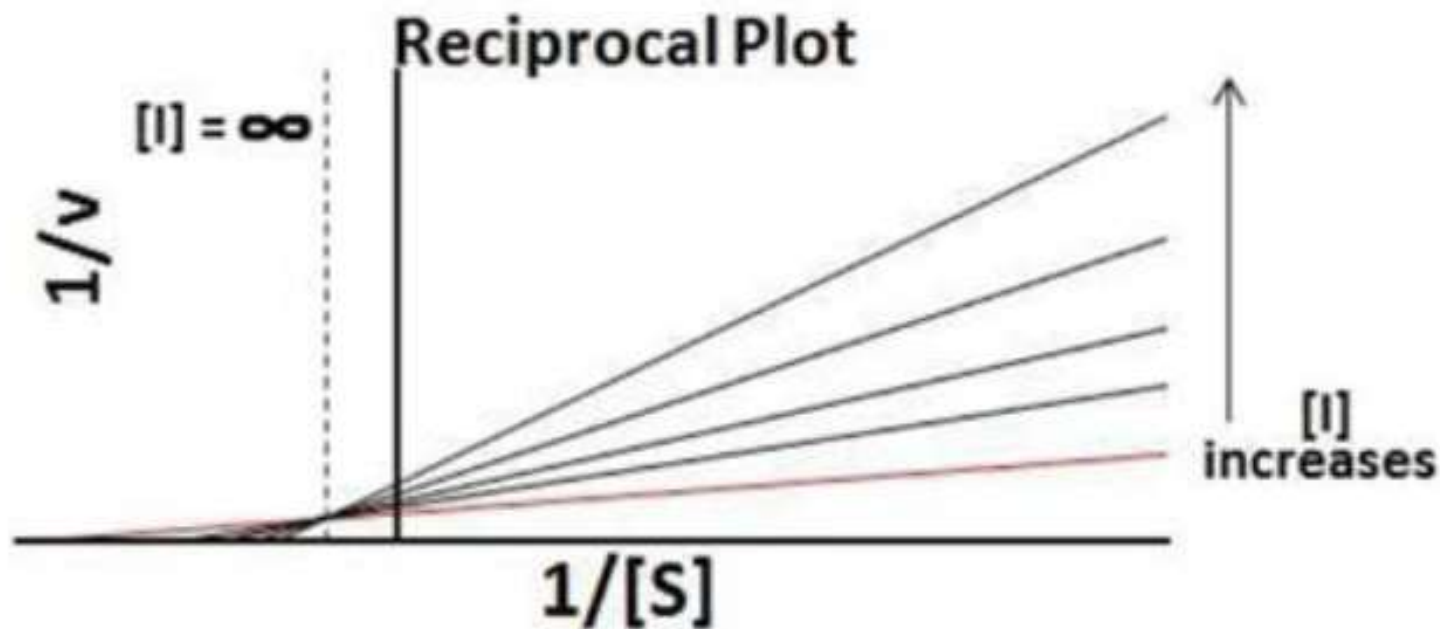


Figure 6. Kinetics of Mixed Inhibition

# ENZYME INHIBITION

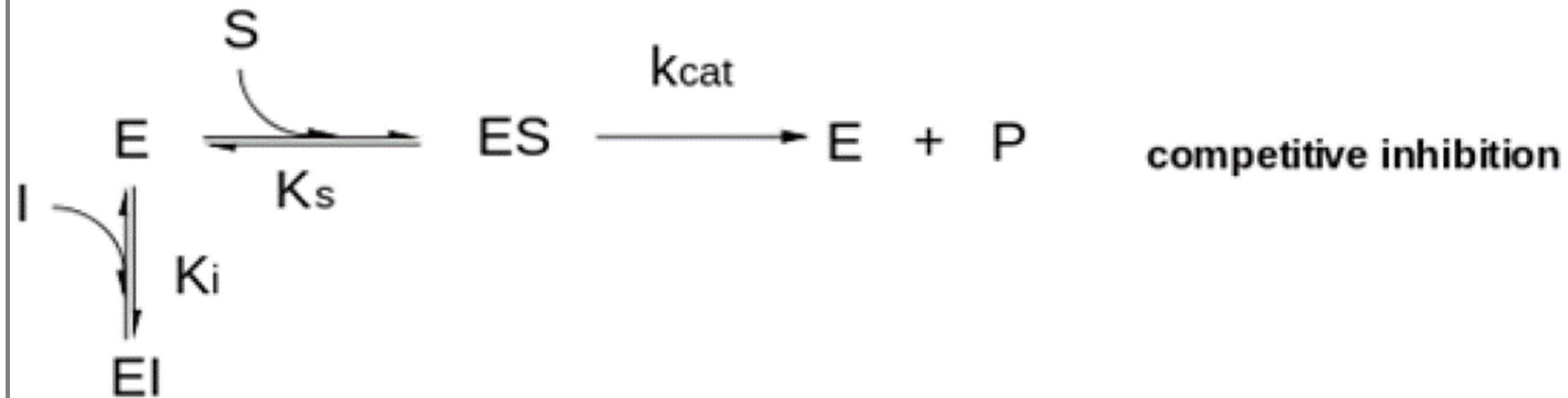


Figure 7. Different types of reversible inhibitions

# ENZYME INHIBITION

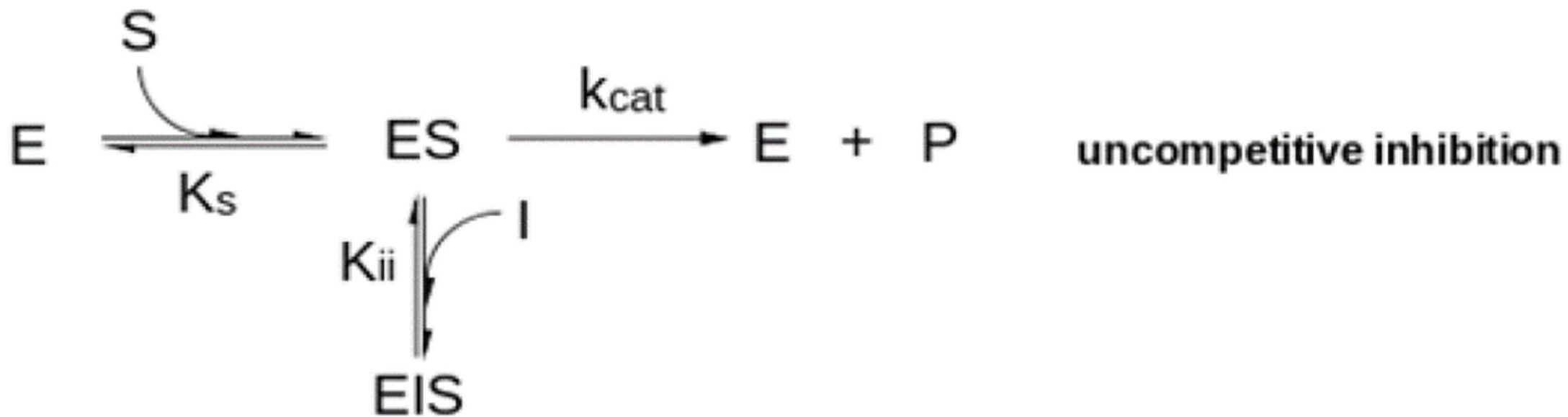


Figure 7. Different types of reversible inhibitions

# ENZYME INHIBITION

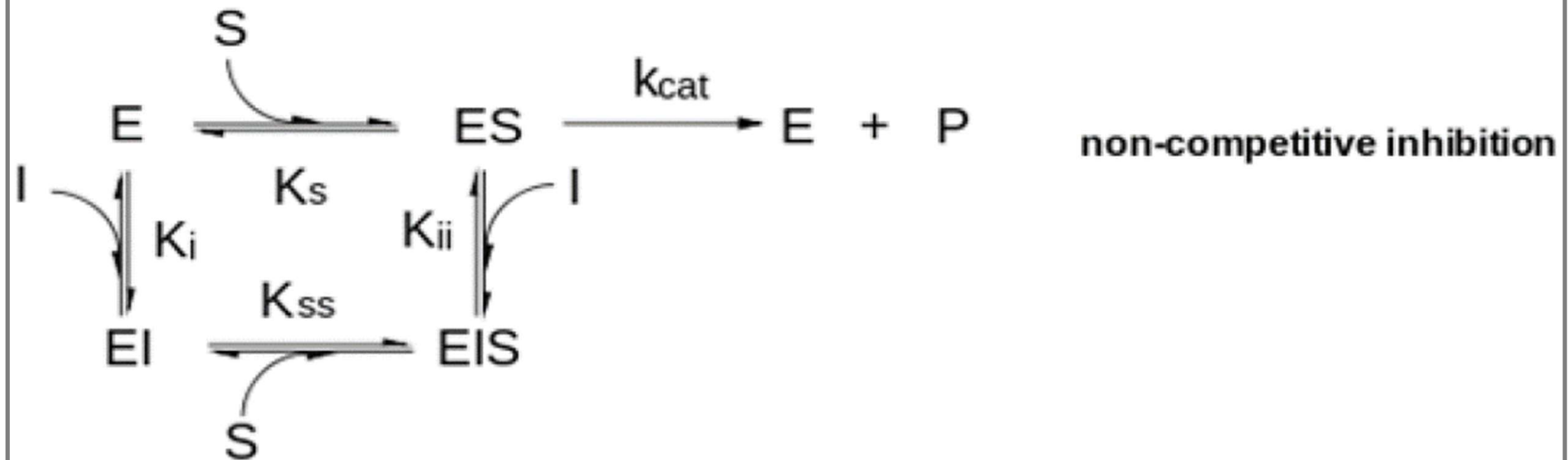


Figure 7. Different types of reversible inhibitions

# ENZYME INHIBITION

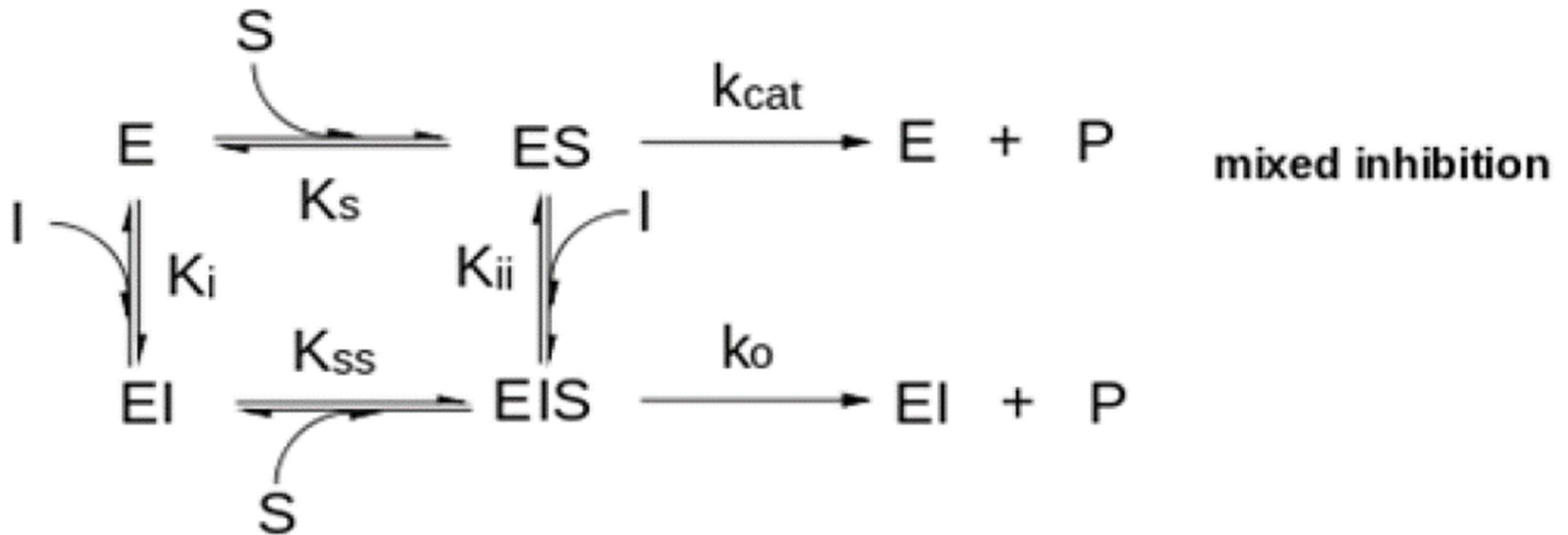


Figure 7. Different types of reversible inhibitions

# ENZYME INHIBITION

## ■ Irreversible Inhibition

In this type of inhibition, the hindrance is of **permanent** nature by *modifying enzyme covalently*. These types of inhibitors often contain **electrophilic functional groups** like **fluorophosphates, aldehydes, haloalkanes, alkenes, nitrogen mustards, phenyl sulfonates, Michael acceptors etc.**, which react with amino acid side chains having *nucleophilic residues*.



# ENZYME INHIBITION

## ■ Irreversible Inhibition

These inhibitors are **very specific** in the mechanism of inactivation for a particular class of enzyme. They do irreversible *inhibition by specially altering the active site*. They display **inhibition which is time-dependent.** ...

# ENZYME INHIBITION

## ■ Irreversible Inhibition

... Their **potency** cannot be characterized by  $IC_{50}$  value. These inhibitors *increase*  $K_m$  and *decrease*  $V_{max}$ .

Example: Di isopropyl fluorophosphate (**DFP**) is an irreversible **protease inhibitor**.

# ENZYME INHIBITION

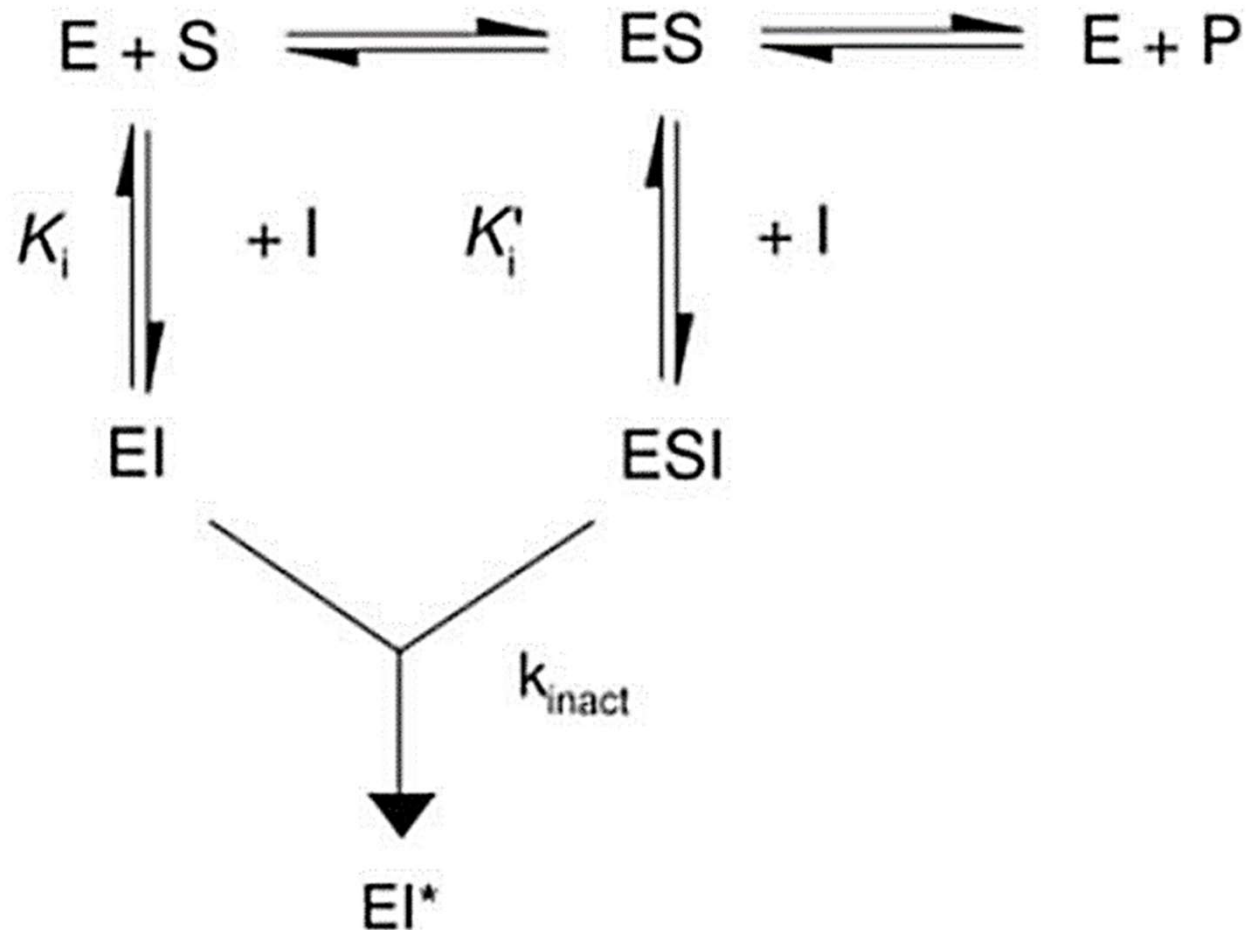


Figure 8. Kinetics of irreversible inhibition

# ENZYME INHIBITION

Table 3. Therapeutic uses of irreversible inhibitors.

Inhibitors	Enzyme inhibited	Therapeutic uses
Disulfiram	Aldehyde dehydrogenase	Treats alcoholism
Cyanide ions	Cytochrome oxidase	Inhibition of the respiratory chain
Fluoride ions	Enolase	Inhibition of Glycolysis
Melathion	Acetyl choline esterase	Used as an organophosphorus insecticide
Di-isopropyl fluorophosphate	Serine proteases, Acetyl choline esterase	Used as a nerve gas
British Anti Lewisite (BAL)	Reaction with the thiol (-SH) group of the enzyme	Used as an antidote in case of poisoning due to heavy metal

# ENZYME INHIBITION

Table 3. Therapeutic uses of irreversible inhibitors.

Inhibitors	Enzyme inhibited	Therapeutic uses
Disulfiram	Aldehyde dehydrogenase	Treats alcoholism
Cyanide ions	Cytochrome oxidase	Inhibition of the respiratory chain
Fluoride ions	Enolase	Inhibition of Glycolysis
Melathion	Acetyl choline esterase	Used as an organophosphorus insecticide
Di-isopropyl fluorophosphate	Serine proteases, Acetyl choline esterase	Used as a nerve gas
British Anti Lewisite (BAL)	Reaction with the thiol (-SH) group of the enzyme	Used as an antidote in case of poisoning due to heavy metal

# ENZYME INHIBITION

## ■ Suicide inhibition

This is another type of *irreversible inhibition*. In this case, the target enzyme **converts** the **inhibitor compound** into a reactive form in its **active site**. They are also known as *mechanism based inhibitors* or *transition state analogs*.

# ENZYME INHIBITION

## ■ Suicide inhibition

Example: **DFMO** [ *$\alpha$ -difluoromethyl ornithine*], an analogue of *ornithine* inhibits *ornithine decarboxylase*.

Example: *Allopurinol* is a suicide inhibitor of *xanthine oxidase*

Example: Aspirin inhibits cyclooxygenase.

# ENZYME INHIBITION

Table 4. Therapeutic uses of suicide inhibitors

Drugs	Product	Target Enzyme	Therapeutic use
Allopurinol	Alloxanthin	Xanthine oxidase	Treatment of gout
5-Fluorouracil	Fluoro-deoxy uridylate	Thymidylate synthase	Cancer treatment
Aspirin	Acetylates the serine residue present in the cyclooxygenase active site	Cyclooxygenase	Used as a non-steroidal anti-inflammatory drug
Difluoro methyl ornithine (DFMO)	Forms an irreversible covalent complex with the co-enzyme	Ornithine decarboxylase	Used for treating Sleeping sickness (trypanosomiasis)



# ENZYME INHIBITION

**Table 5. Differences between reversible & irreversible inhibitions**

Reversible	Irreversible
Binds via non-covalent interactions	Binds via covalent interactions
Do not perform any chemical changes	Inhibitor binds to the substrate and prevent catalytic activity of enzymes
As there is no bonding between the inhibitor and substrate, reversible inhibition can be reversed,	Irreversibility due to strong covalent bonding

# ENZYME INHIBITION

## ■ Allosteric Inhibition

Allosteric inhibition is a type of *enzyme regulation*, in which allosteric inhibitor binds to a site **other than the active site** of the enzyme. This *additional site* to which effector binds is called **allosteric site**. **When** these effectors bind to the **protein**, results with **conformational change** and ...

# ENZYME INHIBITION

## ■ Allosteric Inhibition

... cause *enhancement in activity* is known as *allosteric activators*. When they *decrease the activity* of the protein, they are known as **allosteric inhibitors**. Allosteric enzymes are **K** or **V** types.

# ENZYME INHIBITION

## ▪ Models of Allosteric Regulation

The allosteric effects or **mechanism** is well described by the **concerted MWC model**, which was put forth by *Monod, Wyman and Changeux*. Another model called the **sequential model**, proposed by *Koshland, Nemethy and Filmer*, also possibly explains the allosteric regulation. ...

# ENZYME INHIBITION

## ▪ Models of Allosteric Regulation

... Both these models postulate that **enzyme subunits** exist in one two conformations - **tensed (T) or relaxed(R) states**.

# ENZYME INHIBITION

## ■ Concerted model

This model is known as **symmetry** or **MWC model**. According to this model, **enzyme subunits** exist in same conformation, they **are connected** and a **slight conformational change** in any one of the subunits is conferred to all other subunits of the enzyme. When any **ligand** or **substrate** is absent, the equilibrium favors towards either of the conformational states. Among the **tensed** and **relaxed states**, the **‘R’ state** has **higher affinity than ‘T’ state**. The most successful application of this model is this **regulation of hemoglobin**.

# ENZYME INHIBITION

## ■ Sequential model

In contrary to Concerted model, **Sequential model** states that **enzymes subunits are not connected**, such that any **change** in the **enzyme conformation leads to induction** of a similar change in the others. When a subunit randomly collides with substrate, an **induced fit** converts a subunit from the **‘T’ state** to **‘R’ state**.

# ENZYME INHIBITION

## ■ Morpheein model

The third model is a **dissociative concerted model** known as **Morpheein model**. This is **physiologically significant homo-oligomeric tetramer structure**. Transition in the morpheein model are assisted by dissociation of oligomer, conformational change in dissociated state and reassembly of oligomers.

So far, one of the best characterized morpheein is the enzyme **porphobilinogen synthase**.

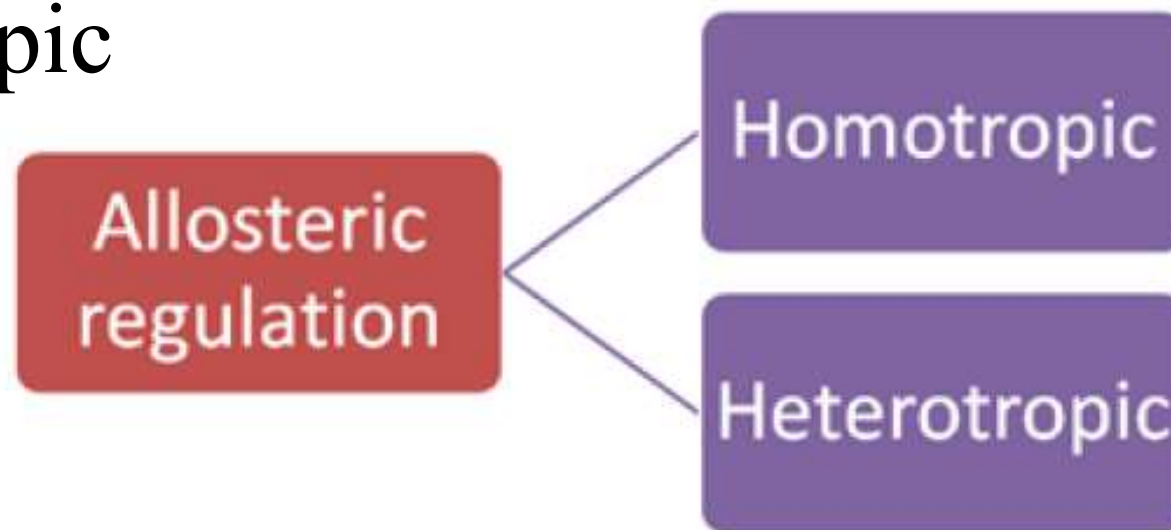


# ENZYME INHIBITION

## ▪ Types of Allosteric Regulation

There are mainly **two types** of allosteric regulation.

- Homotropic
- Heterotropic



# ENZYME INHIBITION

## ▪ Types of Allosteric Regulation

- **Homotropic regulation:** It's a **positive modulation** - The modulator acts not only as a **substrate** but also as a **regulatory molecule** of the target enzyme.

Example:  $O_2$  and  $CO$  are homotropic allosteric modulator of haemoglobin.

- **Heterotropic regulation:** It can either be a **positive or negative modulation**. Here the modulator is a **regulatory molecule** but **not an enzymes substrate**.

Example:  $CO_2$  is a heterotropic modulator of haemoglobin.

# ENZYME INHIBITION

- **Importance of enzyme inhibition**
- Understanding regulation of enzyme activity in living cells.
- Elucidation of the cellular metabolic pathways by accumulation of intermediates.
- Helps in identification of catalytic or functional groups present at the enzyme active site.

# ENZYME INHIBITION

- **Importance of enzyme inhibition**
- Helps in providing information on enzyme's substrate specificity.
- Helps in studying the mechanism of catalytic activity.
- Competitive or suicide inhibitors also find therapeutic applications.

# *Thank you!*



 [www.DiligentEdu.in](http://www.DiligentEdu.in)

[www.YouTube.com/@DiligentEdu](http://www.YouTube.com/@DiligentEdu)

The logo for DiligentEdu features the word "Diligent" in dark blue and "Edu" in orange, both underlined. A small globe icon is placed within the letter 'i' of "Diligent", and a sunburst icon is placed to the left of the 'D'. A copyright symbol (©) is located to the right of the word "Edu".